Natural product syntheses based on asymmetric Diels–Alder reactions

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<u>Abstract</u> - (R)-Pantolactone, (S)-ethyl lactate and other α -hydroxy-carboxylic acid derivatives are effective chiral auxiliaries for large-scale asymmetric Diels-Alder additions of enoates. Mechanistic aspects and preparations of intermediates for EPC-syntheses of carbocyclic nucleoside analogs, prostaglandins and other biologically active compounds are described.

INTRODUCTION

Excellent progress has recently been achieved in the development of chiral auxiliaries for asymmetric Diels-Alder (ADA) reactions (ref. 1). Work of this group has shown that enoates of α -hydroxycarboxylic acid derivatives are very useful chiral dienophiles. (R)- and (S)-lactates 1 and 2 (ref. 2), (R)-pantolactone (3) (ref. 3) and (S)-N-methyl-2-hydroxysuccinimide (4) were found to be particularly effective auxiliaries.



1-3 are large-scale commodities. The imide 4 is available in one step from (S)-malic acid. The combination of low cost, ease of enoate formation and hydrolysis, and ready cristallization of derivatives of 3 and 4 allows practical large-scale preparations (>100 g) of numerous building blocks useful for EPC-syntheses of biologically active compounds.

LEWIS ACID CATALYZED REACTIONS—FUNDAMENTALS

Our mechanistic rationale, inter alia based on crystal structures of enoate/Lewis acid complexes (refs. 4.5), for description of titanium tetrachloride catalyzed reactions of enoates of the auxiliaries 3 and 4 is presented in Scheme 1 for the special case of cyclic dienes (ref. 6). Essential features of the reactive chelate complexes 5 and 6 are syn-enoate conformations and diastereoface-selective shielding of the enoate group by the TiCl₄ moiety. Quite generally, Diels-Alder adducts were obtained with diastereoselectivity of ca. 97:3 (7:8 or 10:9) and pure major isomers resulted after recrystallization.

Scheme 1



Earlier work in the area of ADA reactions was aimed at high degrees of stereoselectivity. Today, high reactivity of the dienophile and practicality are increasingly recognized as equally important features. Touchstones for assessing these aspects are the reactions of transcrotonates with cyclopentadiene and of acrylates with cyclohexadiene, combinations of low reactivity. Carried out according to Scheme 1, no difficulties were encountered (ref. 7).



Some of the carboxylic acids obtained by saponification (LiOH,THF/H₂O) of adducts 7 or monocyclic analogs are shown above. β -Bromoacrylates (R=Br, Scheme 1) are saponified with concomittant elimination to give 14. Similarly, treatment with NaOBn/BnOH yields the β -addition product 15. All these compounds are useful for EPC-syntheses of natural products (ref. 8).

CARBOCYCLIC NUCLEOSIDE ANALOGS

Some natural and unnatural carbocyclic analogs of nucleosides display interesting physiological effects (antiviral, antibiotic, and antitumor activities) (ref. 9). Recent work of our group has been directed at synthesizing compounds of this class via enantiomerically pure Diels-Alder adducts. Here we report a short synthesis of Ohno's lactone (19) which was previously transformed into (-)-aristeromycin (18) (antibiotic) (ref. 10). In the course of this work, an intermediate suitable for a synthesis of the antiviral agent cyclaradine (20) (ref. 11) was also obtained.



18 (-) - aristeromycin

19 Ohno's lactone

20 cyclaradine

(+)(R)-5-Norbornene-2-carboxylic acid (ent-11) yields (1R)-5-norbornen-2-one (21) via a simple three-step sequence (ref. 12)(Scheme 2). Our synthetic plan called for cis-dihydroxylation of the double bond and subsequent oxidative cleavage of the C2-C3-bond. To our surprise, reaction of 21 with osmium tetroxide/N-methylmorpholine N-oxide gave mainly the transaddition product 22t (ratio 22t:22c ca. 4:1). The source of this anomaly became clear when it was found that 22c rearranges to 22t upon treatment with base, obviously via retro-aldol reaction. As a consequence, the hydroxylation was carried out with one equiv. of p-toluenesulfonic acid added to the standard agent. This procedure indeed furnished 22c in quantitative yield. Treatment of 22c with methylamine in methanol gave 22t in high yield. Transformation of 22c into Ohno's lactone, via the acetal 23 and ozonization of its silyl enol ether, was uneventful. The total yield of 19 from 5-norbornene-2-carboxylic acid was 54 %. A route from 22t to cyclaradine (20) is currently being worked out.



NON-CATALYZED AND LEWIS ACID CATALYZED REACTIONS OF FUMARATES

In view of their potential experimental simplicity and broad range of applications, non-catalyzed ADA additions are of great interest. However, until very recently little success was achieved with esters. In contrast, the acrylate 24 of (S)-ethyl lactate (2) reacts diastereoselectively with cyclopentadiene at 0 $^{\circ}$ C in hexane (endo-add.: 80:20, exo-add.: 85:15) (ref. 2). The (typically) low endo-exo ratio of 1.7 prevents direct preparative use of these results. But it was of interest to examine corresponding fumarates which cannot give rise to endo-exo isomers and, furthermore, were expected to show synergistic activity of their auxiliary groups and, in consequence, a higher level of selectivity than acrylates (cf. refs. 1a, 4). The fumarate 26, displaying diastereoselectivity of 98:2 with cyclopentadiene in n-hexane (Scheme 3), is indeed the more selective dienophile. It is assumed that the reaction proceeds via the C_{2} -symmetric species 26 with anti-enoate conformation (cf. Formula) for which attack from the more accessible back-face is favored.



For TiCl₄-catalyzed reactions of the acrylate 24, which yield preferred products of opposite configuration compared to the non-catalyzed reaction, the complex 25 with syn-enoate conformation is postulated as reactive species (cf. Scheme 1). On extending this model to the fumarate 26 one would expect the 1:1 complex 27 with syn- and anti-enoate conformations as reactive species, and therefore antagonistic action of the auxiliary groups, and/or the 2:1 complex 28 with syn-enoate conformations on both sides, and therefore synergistic action of the auxiliary groups. For differentiation of these complexes, 26 was allowed to react with cyclopentadiene using various TiCl₄: 26 ratios and the products (Scheme 3) were analyzed by HPLC. We interpret the results of this series of experiments (Fig. 1) as follows: 27 determines the mode of reaction at low TiCl₄: 26 ratios (A), hence the non-selective reaction. In the range B, competition of the complex 28 becomes noticeable, which in range C finally dominates and yields the ester 29b with 95:5 selectivity (ref. 4).

Results obtained with the fumarate of (R)-pantolactone, **30**, are displayed in Scheme 4. Compared to **26**, the fumarate **30** yields Diels-Alder adducts of opposite configuration due to inverse chirality sense of the HO-C-CO moieties of 2 and 3. Thus, for an extensive variety of chiral trans-dicarboxylic acids both enantiomers are accessible via non-catalyzed reactions. Lewis acid promoted reactions of the fumarate **30** also proceed with high selectivity when EtAlCl₂ is employed in excess. For a reason not yet understood, in conjunction with **30**, EtAlCl₂ is more effective than TiCl₄.

Diels-Alder reactions with the inexpensive fumarates 26 and 30 are of interest for EPC-syntheses of physiologically active compounds - e.g., the (2S,3S)-5-norbornene-2.3-dicarboxylic acid readily obtainable (ref. 1a) from adducts 29b or 31a is a starting material for the ICI prostaglandin synthesis (ref. 13). Another, less obvious entry into the prostaglandin series is shown in Scheme 5. Key step is the decarboxylative 1.3-elimination of the easily prepared iodolactone 35 which furnishes the nortricyclene 36 in quantitative yield. From the lactone 36, straightforward synthetic tactics leads to the Corey-Sutherland intermediate 39 (ref. 13) and the syn-isomer 38 (76 % yield from 35) which opens avenues to a large variety of important natural products (cf. ref. 14).



Fig. 1. Dependence of the diastereoselectivity of the reaction of 26 with cyclopentadiene on the ratio $TiCl_a$:26.



Scheme 4



Non-catalyzed Solvent: carbon tetrachloride					Catalyzed: 4 equiv. of EtAlCl Solvent: methylene chloride		
Diene	Temp. (,C1	a:b ^b	Y.[%]	Temp. [^O C]	a:b	Y. [%] ^C
\bigcirc	0	94 cr >>99	.5:5.5 :1	98 80	30	7.5:92.5 cr << 1 :99	90 60
\bigcirc	80	88 cr >>99	:12 : 1	95 61	-24	3 :97 cr << 1 :99	90 60
	80	70 cr >>99	:30 :1	100 60	24	3 :97 cr << 1 :99	90 50
	110 ^a	97 cr >>99	: 3 : 1	95 82	0	4 :96 cr << 1 :99	80 61

(a) Solvent: toluene; (b) cr: after crystallization; (c) yields are not optimized.

Scheme 5



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