Studies on the biosynthesis of antibiotics

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<u>Abstract</u> - Studies with stable isotope-labeled precursors followed by NMR analysis of the resulting antibiotics have revealed the biosynthetic origin and mode of formation of the cyclohexanecarboxylic acid moiety in ansatrienin, the mC_7N unit in ansatrienin and asukamycin, the 2-amino-3-hydroxycyclopent-2-enone moiety of asukamycin and reductiomycin, and the dihydrofuranylacrylic acid moiety of reductiomycin.

This report deals with the biosynthesis of three antibiotics, asukamycin (1) from *Streptomyces nodosus* var. *asukaensis* (ref. 1), ansatrienin (mycotrienin) (2, Figure 2) from *S. collinus* and *S. rishiriensis*



(ref. 2.3) and reductiomycin (3. Figure 4) from S. xanthochromogenus (ref. 4). These compounds contain several biosynthetically unique structural units, namely a cyclohexane ring (1 and 2). a mC_7N unit (1 and 2) and a 2-amino-3-hydroxycyclopent-2-enone moiety (C_5N unit) (1 and 3). In 1 and 2 these are either connected by or attached to a framework of olefinic or saturated carbon chains which are derived biosynthetically from acetate/propionate units via the polyketide pathway.

The cyclohexanecarboxylic acid moiety of **2** was found to arise intact from the seven carbon atoms of shikimic acid (**4**. Figure 2) via 1-cyclohexenecarboxylic acid (ref. 5). To obtain more information on the mode of conversion of **4**, a precursor of aromatic compounds. into a fully reduced hydroaromatic ring, we synthesized $[2^{-13}C]$ -**4** and examined

its incorporation into 2. The ¹³C-NMR spectrum of the resulting sample of 2 showed only one enriched carbon, C-2 or C-6 of the cyclohexane ring. By degradation to the S-mandelate ester of cyclohexylcarbinol and NMR comparison with independently synthesized samples of the R- and Smandelate esters of [1R. 2R)-[2-2H,]cyclohexylcarbinol it was established that the label resided exclusively at C-6, i.e., the processing of $[2^{-13}C]$ -4 was stereospecific to give 2 with S configuration at C-1 of the cyclohexane ring (ref. 6). Similarly, $[2^{-2}H_1]$ -4 labeled 2 exclusively in the pro-6R (axial) position of the cyclohexane ring (ref. 7). A similar conclusion had been drawn, from a feeding experiment with [6-2H,]glucose, by Furukawa et al. (ref. 8) on the biosynthesis of the cyclohexane ring in w-cyclohexylundecanoic acid. However, their interpretation that this represents a Re, Re reduction of the double bond in 4 is contradicted by our observation that a small amount of the 1cyclohexenecarboxylic acid analog accompanying the sample of $2 \text{ from } [2^{-13}C]-4$ carried the ¹³C label at C-6 (CH₂) rather than at C-2 (CH), i.e., the double bond migrates in the ring. In addition, 6deuterated 4 gave no deuterium incorporation into 2 (ref. 7). Based on these results we propose the pathway shown in Figure 1 for the formation of the cylcohexanecarboxylic acid moiety of 2. Attempts to establish the mode of formation of the cyclohexane ring of $\mathbf{1}$ in the same way met with failure because of very poor incorporation of 4, presumably due to permeability problems. However, the coupling patterns in the cyclohexane ring and adjacent carbon of 1 biosynthesized from [U-¹³C₂]glycerol leave no doubt about its shikimate origin (ref. 9).



A number of antibiotics, mainly ansamycins and mitomycins, contain mC_7N units, 6-membered carbocyclic rings carrying a carbon and a nitrogen substituent in a 1.3 (meta) orientation. Their origin has been traced to the shikimate pathway, but 4 itself is not incorporated into these mC_7N units. This could mean that either 4 is not taken up into the producing organism or the formation of the mC_7N unit branches off earlier in the shikimate pathway. 3-Amino-5-hydroxybenzoic acid (AHBA) has been established as a specific precursor of these mC_7N units (ref. 10, 11). Consistent with this we observed very efficient incorporation of $[7-^{13}C]AHBA$ into 2, labeling exclusively C-17 of the antibiotic (Figure 2). Since the feeding of $[2-^{13}C]-4$ described earlier had labeled the cyclohexane ring of 2 but not the mC_7N unit, we can conclude that, at least in this case, non-incorporation of 4 is not due to permeability problems. Hence, the formation of AHBA must branch off earlier in the shikimate



pathway. Analysis of the ${}^{13}C_{-}{}^{13}C$ coupling pattern of a sample of 2 biosynthesized from [U- ${}^{13}C_{3}$]glycerol revealed that the nitrogen of the mC₇N unit is attached to the carbon corresponding to C-5, not C-3, of 4 (Figure 2) (ref. 12), consistent with the observations of Hornemann *et al.* on mitomycin (ref. 13) and Rinehart *et al.* on geldanamycin (ref. 14) biosynthesis. Based on these results and the finding of Jiao *et al.* (ref. 15) that the amide nitrogen of glutamine is the best source of the nitrogen of rifamycin, we propose the pathway shown in Figure 3 for the formation of AHBA, the precursor of the mC₇N unit.

Surprisingly, a feeding experiment with $[7-^{13}C]AHBA$ gave no incorporation into 1 at all (ref. 9). Further examination revealed a completely different pathway for the formation of the mC₇N unit in 1 involving succinic acid (carbon atoms 6, 5, 4 and 7) and glycerol (carbons 1, 2, 3) as precursor. Details of this new pathway need to be established.



Inspection of the $C_5 N$ unit in 1 and 3 shows that it contains all the carbon, hydrogen, oxygen and nitrogen atoms of 5-aminolevulinic acid (ALA) less one molecule of water. In a series of feeding experiments with glycine and ALA it was established that the $C_5 N$ unit in both 1 and 3 does indeed arise by an intramolecular cyclization of ALA (ref. 16, 17). The most conclusive evidence comes from a feeding experiment with [4.5-¹³C2]ALA that gave 3 in which one half of the labeled molecules showed coupling between C-1 and C-2 and the other half between C-2 and C-3 of the $C_5 N$ unit (Figure 4) (ref. 17). It is proposed that the cyclization of ALA is mediated by an enzyme using pyridoxal phosphate as a cofactor.

The remaining 9 carbon atoms of 3 where first thought to arise from tyosine. However, it was quickly found that the acetoxy group comes from acetate, leaving only seven carbon atoms to account for. Coupling analysis of 3 biosynthesized from $[U^{-13}C_3]$ glycerol pointed to a pathway involving ring cleavage of a symmetrical precursor generated via the shikimate pathway (ref. 17). Following this lead it was found that p-hydroxy- $[7^{-13}C]$ benzoic acid and p-hydroxy- $[7^{-13}C]$ benzaldehyde efficiently and specifically labeled 3 (55-64% and 35% enrichment, respectively at C-5') (ref. 17, 18). Of the four ring hydrogens of p-hydroxy- $[2.3,5.6^{-2}H_4]$ benzoic acid three are completely retained in the conversion and only the one appearing at C-2'' of 3 undergoes partial exchange with solvent protons, possibly during a double bond *cis-trans* isomerization (ref. 18). The pathway of 3 biosynthesis can thus be formulated as shown in Figure 4.



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