Table 16: Hg^{2+}-Selective Electrodes

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
Hg^{2+}-1	**Hg^{2+}-1** (w = 1–4 %), DDP (w = 66–69 %), PVC (w = 30 %)	Ca^{2+}, –1.8; Co^{2+}, –1.0; Ni^{2+}, –0.7; Zn^{2+}, –1.1; Cd^{2+}, –1.4; Pb^{2+}, –1.3; Hg^{2+}, +1.0	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –1.6; Co^{2+}, –0.5; Ni^{2+}, –0.5; Zn^{2+}, –1.1; Cd^{2+}, –0.8; Pb^{2+}, 0.0; Hg^{2+}, +1.0	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$; pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	
		Ca^{2+}, –1.1; Co^{2+}, –0.7; Ni^{2+}, –0.3; Zn^{2+}, –1.1; Cd^{2+}, –0.7; Pb^{2+}, 0.0; Hg^{2+}, +3.6	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
	Hg^{2+}-1 (w = 1–4 %), DDP (w = 66–69 %), PVC (w = 30 %), KTpClPB (x_i = 70 %)	Ca^{2+}, –0.2; Co^{2+}, –0.7; Ni^{2+}, –0.3; Zn^{2+}, –0.8; Cd^{2+}, +0.3; Pb^{2+}, +0.6; Hg^{2+}, +7.8	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –0.3; Co^{2+}, –0.7; Ni^{2+}, –0.3; Cd^{2+}, –0.8; Pb^{2+}, +0.2; Hg^{2+}, +6.0	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	

Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
		Ca^{2+}, –0.8; Co^{2+}, –0.3; Ni^{2+}, –0.1; Zn^{2+}, –0.5; Cd^{2+}, –0.3; Pb^{2+}, +0.1; Hg^{2+}, +4.0	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
	Hg^{2+}-1 (w = 1 %), oNPOE (w = 69 %), PVC (w = 30 %)	Na^+, –1.2; Ca^{2+}, –1.4; Co^{2+}, –1.0; Ni^{2+}, –1.2 Cu^{2+}, –0.9; Zn^{2+}, –2.4; Cd^{2+}, –2.0; Pb^{2+}, –1.8; Ag^+, +1.9	SSM	0.01	0.01	–	–	conditioned overnight in H_2O; internal electrolyte, 10^{-2} M $CuCl_2$ pH = 3	[2]
		Na^+, –4.3; Ca^{2+}, –2.9; Ni^{2+}, –2.6; Cu^{2+}, –2.4; Zn^{2+}, –2.7; Cd^{2+}, –2.9; Pb^{2+}, –2.7; Ag^+, +2.2	SSM	0.01	0.01	–	–	conditioned in 10^{-3} M $HgCl_2$ for 2 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –4.5; Ca^{2+}, –3.3; Ni^{2+}, –2.9; Cu^{2+}, –2.6; Zn^{2+}, –3.1; Cd^{2+}, –3.1; Pb^{2+}, –2.9; Ag^+, +2.3	SSM	0.01	0.01	–	–	conditioned in 10^{-3} M $HgCl_2$ for 6 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –4.0; Ca^{2+}, –3.1; Ni^{2+}, –2.9; Cu^{2+}, –2.7; Zn^{2+}, –2.6; Cd^{2+}, –2.6; Pb^{2+}, –2.9; Ag^+, +2.3	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$ for 40 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –3.7; Ca^{2+}, –2.7; Ni^{2+}, –2.9; Cu^{2+}, –2.7; Zn^{2+}, –2.9; Cd^{2+}, –2.9; Pb^{2+}, –2.7; Ag^+, +1.8	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$, pH = 3; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
		Na^+, –0.1; Ca^{2+}, –1.6; Ni^{2+}, –1.8; Cu^{2+}, –2.7; Zn^{2+}, –1.8; Cd^{2+}, –2.2; Pb^{2+}, –1.9; Ag^+, +3.0	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M KCl, pH = 3; internal electrolyte, 10^{-2} M KCl, pH = 3	

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Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
		Na^+, –1.3; Ca^{2+}, –1.7; Ni^{2+}, –2.4; Zn^{2+}, –2.4; Cd^{2+}, –2.0; Pb^{2+}, –1.7; Ag^+, +2.4	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
	Hg^{2+}-1 (w = 1 %), oNPOE (w = 66–69 %), KTpClPB (x_i = 70 %), PVC (w = 30 %)	Na^+, –4.4; Co^{2+}, –4.8; Ni^{2+}, –5.6; Zn^{2+}, –5.8; Pb^{2+}, –3.6; Ag^+, +1.6	SSM	0.01	0.01	41	10^{-5}–10^{-3}	conditioned overnight in H_2O; t_{resp} < 45 s; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[2]
		Na^+, –5.0; Ca^{2+}, –4.2; Ni^{2+}, –3.7; Cu^{2+}, –3.5; Zn^{2+}, –3.8; Cd^{2+}, –3.6; Pb^{2+}, –3.7; Ag^+, +2.4	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$, for 2 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –5.2; Ca^{2+}, –4.5; Ni^{2+}, –4.2; Cu^{2+}, –3.5; Zn^{2+}, –4.0; Cd^{2+}, –3.9; Pb^{2+}, –3.9; Ag^+, +2.3	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$, for 6 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –4.6; Ca^{2+}, –4.0; Ni^{2+}, –3.5; Cu^{2+}, –3.0; Zn^{2+}, –3.2; Cd^{2+}, –3.0; Pb^{2+}, –3.6; Ag^+, +2.2	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$, for 40 d, pH = 2; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 2	
		Na^+, –5.4; Ca^{2+}, –2.7; Ni^{2+}, –3.9; Zn^{2+}, –3.9; Cd^{2+}, –3.9; Pb^{2+}, –3.7; Ag^+, +2.6	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $HgCl_2$, pH = 3; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
		Na^+, +1.3; Ca^{2+}, –0.8; Ni^{2+}, –0.9; Cu^{2+}, –0.6; Zn^{2+}, –0.9; Cd^{2+}, –1.3; Pb^{2+}, –1.0; Ag^+, +2.8	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M KCl, pH = 3; internal electrolyte, 10^{-2} M KCl, pH = 2	
		Na^+, –2.0; Ca^{2+}, –3.7; Ni^{2+}, –3.4; Zn^{2+}, –3.4 Cd^{2+}, –4.0; Pb^{2+}, –3.7; Ag^+, +1.3	SSM	0.01	0.01	–	–	conditioned in 10^{-3}M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 2	

Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
Hg^{2+}-2	**Hg^{2+}-2** (w = 1–4 %), DDP (w = 66–69 %), PVC (w = 30 %)	Ca^{2+}, –1.1; Co^{2+}, –0.5; Ni^{2+}, –0.5; Zn^{2+}, –0.7; Cd^{2+}, +0.3; Pb^{2+}, –0.3; Hg^{2+}, +0.3	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 2	[1]
		Ca^{2+}, –1.6; Co^{2+}, –0.8; Ni^{2+}, –0.4; Zn^{2+}, –0.9; Cd^{2+}, –1.2; Pb^{2+}, –0.7; Hg^{2+}, +1.3	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	
		Ca^{2+}, –1.3; Co^{2+}, –0.7; Ni^{2+}, –0.2; Zn^{2+}, –1.2; Cd^{2+}, –0.5; Pb^{2+}, +0.3; Hg^{2+}, +4.4	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
	Hg^{2+}-2 (w = 1–4 %), DDP (w = 66–69 %), PVC (w = 30 %), KTpClPB (x_i = 70 %)	Ca^{2+}, +0.5; Co^{2+}, +0.2; Ni^{2+}, +0.1; Zn^{2+}, +0.1; Cd^{2+}, +0.3; Pb^{2+}, +0.2; Hg^{2+}, +5.8	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte: 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –0.4; Co^{2+}, –0.2; Ni^{2+}, –0.1; Zn^{2+}, –0.5; Cd^{2+}, +0.1; Pb^{2+}, +0.0; Hg^{2+}, +5.6	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	

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Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
		Ca^{2+}, –1.0; Co^{2+}, –0.6; Ni^{2+}, –0.4; Zn^{2+}, –1.3; Cd^{2+}, –0.8; Pb^{2+}, –0.4; Hg^{2+}, +3.8	FIM	–	0.01	–	–	*K* was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $Cu(NO_3)_2$, pH 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	
	Hg^{2+}-2 (*w* = 1 %), oNPOE (*w* = 69 %), PVC (*w* = 30 %)	Na^+, –1.5; Ca^{2+}, –2.2; Ni^{2+}, –1.7; Zn^{2+}, –1.8 ; Cd^{2+}, –2.4; Pb^{2+}, –2.2; Ag^+, +1.3	SSM	0.01	0.01	–	–	conditioned overnight in H_2O; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[2]
	Hg^{2+}-2 (*w* = 1 %), oNPOE (*w* = 69 %), PVC (*w* = 30 %), KTpClPB (x_i = 70 %)	Na^+, –2.7; Ca^{2+}, –4.1; Ni^{2+}, –4.2; Co^{2+}, –4.0; Zn^{2+}, –4.5; Cd^{2+}, –4.8; Pb^{2+}, –4.2; Ag^+, +1.6	SSM	0.01	0.01	38	10^{-5}–10^{-3}	conditioned overnight in H_2O; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[2]
Hg^{2+}-3	**Hg^{2+}-3** (*w* = 1–4 %), DDP (*w* = 66–69 %), PVC (*w* = 30 %)	Ca^{2+}, –2.0; Co^{2+}, +0.3; Ni^{2+}, –1.1; Zn^{2+}, –1.0; Cd^{2+}, +1.3; Pb^{2+}, +0.3; Hg^{2+}, +1.2	FIM	–	0.01	–	–	*K* was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –1.0; Co^{2+}, –0.3; Ni^{2+}, –0.2; Zn^{2+}, –0.4; Cd^{2+}, +0.1; Pb^{2+}, +0.4; Hg^{2+}, +0.6	FIM	–	0.01	–	–	*K* was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –1.7; Co^{2+}, –0.7; Ni^{2+}, –0.4; Cd^{2+}, –0.5; Pb^{2+}, –0.3; Hg^{2+}, +2.0	FIM	–	0.01	–	–	*K* was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	[1]

Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
	Hg^{2+}-3 (w = 1–4 %), DDP (w = 66–69 %), PVC (w = 30 %), KTpClPB (x_i = 70 %)	Ca^{2+}, –0.7; Co^{2+}, –0.3; Ni^{2+}, –0.1; Zn^{2+}, –0.1; Cd^{2+}, +0.6; Pb^{2+}, +0.5; Hg^{2+}, +3.3	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned overnight in 10^{-3} M $CuCl_2$, pH = 3; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, –1.0; Co^{2+}, –0.4; Ni^{2+}, –0.4; Zn^{2+}, –1.1; Cd^{2+}, –0.8; Pb^{2+}, –1.7; Hg^{2+}, +3.0	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 3 d in 10^{-3} M $Cu(NO_3)_2$, pH = 4; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[1]
		Ca^{2+}, +0.4; Co^{2+}, +1.3; Ni^{2+}, +0.2; Zn^{2+}, +1.1; Cd^{2+}, +1.4; Pb^{2+}, +1.5; Hg^{2+}, +4.4	FIM	–	0.01	–	–	K was obtained as $\lg K_{Cu^{2+},B^{n+}}$.; conditioned for 2 weeks in 10^{-3} M $CuNO_3$, pH = 4; internal electrolyte, 10^{-2} M $HgCl_2$, pH = 3	[1]
	Hg^{2+}-3 (w = 1 %), oNPOE (w = 69 %), PVC (w = 30 %)	Na^+, –1.0; Ca^{2+}, –1.8; Ni^{2+}, –1.1; Cu^{2+}, –1.3; Zn^{2+}, –0.9; Cd^{2+}, –2.1; Pb^{2+}, –1.8; Ag^+, +1.6	SSM	0.01	0.01	–	–	conditioned overnight in H_2O; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[2]
	Hg^{2+}-3 (w = 1 %), oNPOE (w = 69 %), PVC (w = 30 %), KTpClPB (x_i = 70 %)	Na^+, +0.4; Ca^{2+}, –1.7; Ni^{2+}, –1.1; Cu^{2+}, –1.4; Zn^{2+}, –1.9; Cd^{2+}, –2.1; Pb^{2+}, –1.7; Ag^+, +1.7	SSM	0.01	0.01	–	–	conditioned overnight in H_2O; internal electrolyte, 10^{-2} M $CuCl_2$, pH = 3	[2]
Hg^{2+}-4	**Hg^{2+}-4** (w = 1 %), DOP (w = 20–50 %), PVC (w = 80–49 %)	Co^{2+}, –2.06; Ni^{2+}, –2.60; Cu^{2+}, –1.15; Cd^{2+}, –2.35; Pb^{2+}, –0.77; Bi^{3+}, +0.11; Fe^{3+}, +0.70; Ce^{3+}, –1.66	MSM	0.01	0.01	27	10^{-5}–10^{-2}	coated graphite elec.; pH = 3.4	[3]
Hg^{2+}-5	**Hg^{2+}-5** (w = 2 %), oNPOE (w = 66 %),	Li^+, –3.0; Na^+, –2.9; K^+, –2.8; NH_4^+, –2.8;	SSM	10^{-3}	10^{-3}	–	–	pH = 4.5	[4]

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Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
	PVC (w = 32 %), KTpClPB (x_i = 5 %)	Mg^{2+}, –6.0; Ca^{2+}, –5.9; Mn^{2+}, –6.0; Co^{2+}, –6.0; Ni^{2+}, –6.2; Cu^{2+}, –6.1; Zn^{2+}, –6.2; Cd^{2+}, –6.1; Pb^{2+}, –5.7; Cr^{3+}, –7.0; Fe^{3+}, –7.1; Ag^{+}, –0.7							
Hg^{2+}-6	**Hg^{2+}-6** (w = 2 %), oNPOE (w = 66 %), PVC (w = 32 %), KTpClPB (x_i = 5 %)	Li^{+}, –5.8; Na^{+}, –5.8; K^{+}, –5.6; NH_4^{+}, –5.6; Mg^{2+}, –8.7; Ca^{2+}, –8.5; Mn^{2+}, –9.1; Co^{2+}, –8.8; Ni^{2+}, –8.7; Cu^{2+}, –8.2; Zn^{2+}, –9.2; Cd^{2+}, –8.9; Pb^{2+}, –7.9; Cr^{3+}, –10.1; Fe^{3+}, –10.3; Ag^{+}, –2.2	SSM	10^{-3}	10^{-3}	–	–	pH = 4.5	[4]
Hg^{2+}-7	**Hg^{2+}-7** (w = 2 %), oNPOE (w = 66 %), PVC (w = 32 %), KTpClPB (x_i = 5 %)	Li^{+}, –3.7; Na^{+}, –4.1; K^{+}, –3.3; NH_4^{+}, –3.7; Mg^{2+}, –6.8; Ca^{2+}, –6.6; Mn^{2+}, –7.6; Co^{2+}, –7.2; Ni^{2+}, –8.0; Cu^{2+}, –8.2; Zn^{2+}, –7.9; Cd^{2+}, –7.9; Pb^{2+}, –2.3; Cr^{3+}, –8.0; Fe^{3+}, –8.3; Ag^{+}, +0.6	SSM	10^{-3}	10^{-3}	–	–	pH = 4.5	[4]
Hg^{2+}-8	**Hg^{2+}-8** (w = 2 %), oNPOE (w = 66 %), PVC (w = 32 %), KTpClPB (x_i = 5 %)	Li^{+}, –9.0; Na^{+}, –9.1; K^{+}, –8.1; NH_4^{+}, –8.6; Mg^{2+}, –12.2 ; Ca^{2+}, –12.0; Mn^{2+}, –12.0; Co^{2+}, –11.8; Ni^{2+}, –12.0; Cu^{2+}, –12.1; Zn^{2+}, –12.1; Cd^{2+}, –11.8; Pb^{2+}, –6.5; Cr^{3+}, –13.1; Fe^{3+}, –12.8; Ag^{+}, –4.7	SSM	10^{-3}	10^{-3}	ca. 70	10^{-5}–10^{-2}	pH = 4.5; $t_{95} \approx 10$ s	[4]
		Li^{+}, –6.0; Na^{+}, –6.1; K^{+}, –5.5; NH_4^{+}, –5.8; Mg^{2+}, –7.6; Ca^{2+}, –7.5; Mn^{2+}, –7.5; Co^{2+}, –7.4; Ni^{2+}, –7.5; Cu^{2+}, –7.6; Zn^{2+}, –7.6; Cd^{2+}, –7.4; Pb^{2+}, –4.8; Cr^{3+}, –8.1; Fe^{3+}, –7.9; Ag^{+}, –3.9	SSM	10^{-3}	10^{-3}			K values were recalculated using the observed slope value.	

Table 16: Hg^{2+}-Selective Electrodes (*Continued*)

ionophore	membrane composition	$\lg K_{Hg^{2+},B^{n+}}$	method	primary ion conc. (M)	interfering ion conc. (M)	slope (mV/ decade)	linear range (M)	remarks	ref.
Hg^{2+}-9	**Hg^{2+}-9** (w = 2 %), oNPOE (w = 66 %), PVC (w = 32 %), KTpClPB (x_i = 5 %)	Li^+, –4.9; Na^+, –5.0; K^+, –3.3; NH_4^+, –4.0; Mg^{2+}, –8.0; Ca^{2+}, –8.6; Mn^{2+}, –8.3; Co^{2+}, –7.7; Ni^{2+}, –8.5; Cu^{2+}, –8.7; Zn^{2+}, –8.3; Cd^{2+}, –8.2; Pb^{2+}, –5.0; Cr^{3+}, –9.5; Fe^{3+}, –9.1; Ag^+, –1.6	SSM	10^{-3}	10^{-3}	–	–	pH, 4.5	[4]

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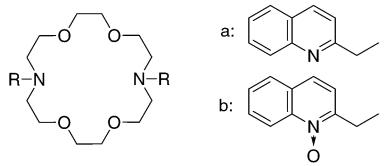
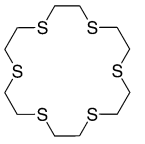


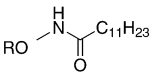
Hg^{2+}-1 (M_r = 572.75): R=a

Hg^{2+}-2 (M_r = 604.75): R=b

Hg^{2+}-3 (M_r = 599.00): R=c

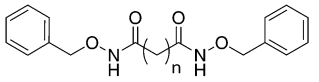


Hg^{2+}-4 (M_r = 360.69)



Hg^{2+}-5 (M_r = 229.37): R=-CH_3

Hg^{2+}-6 (M_r = 305.46): R=-$CH_2C_6H_5$



Hg^{2+}-7 (M_r = 356.43): n=4
Hg^{2+}-8 (M_r = 412.53): n=8
Hg^{2+}-9 (M_r = 482.67): n=13